

Supplementary Data for

Synthesis, Characterization and Applications of (*E*)-3-((5-bromo-2-hydroxy-3-methoxycyclohexa-1,3-dienyl)methyleneamino)-6-(hydroxymethyl)-tetrahydro-2*H*-pyran-2,4,5-triol

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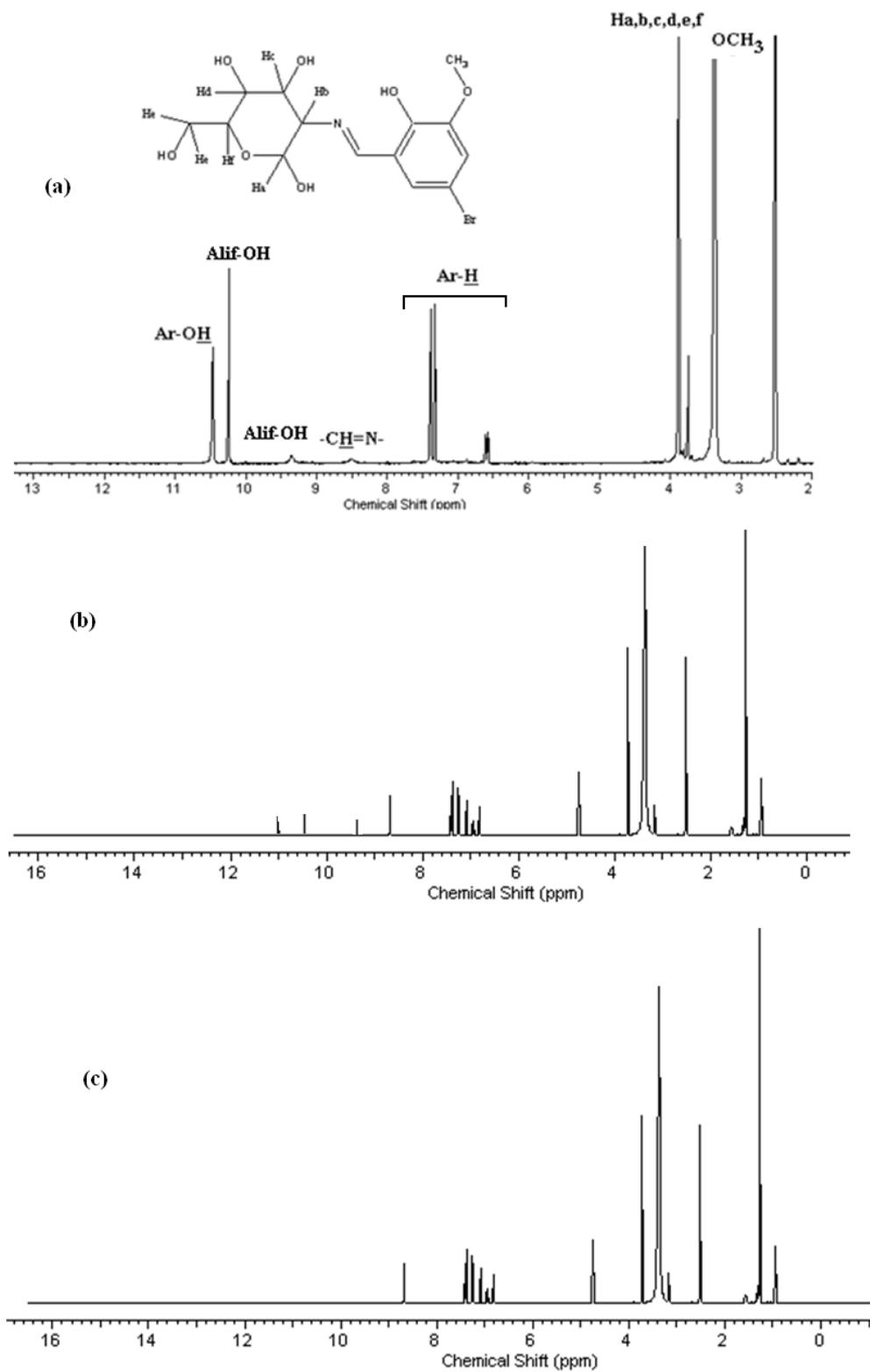


Fig. S1 ^1H -NMR Titration spectrum of the compound. (a) 0.0 eq F^- , (b) 0.5 eq F^- , (c) 3.0 eq F^- .

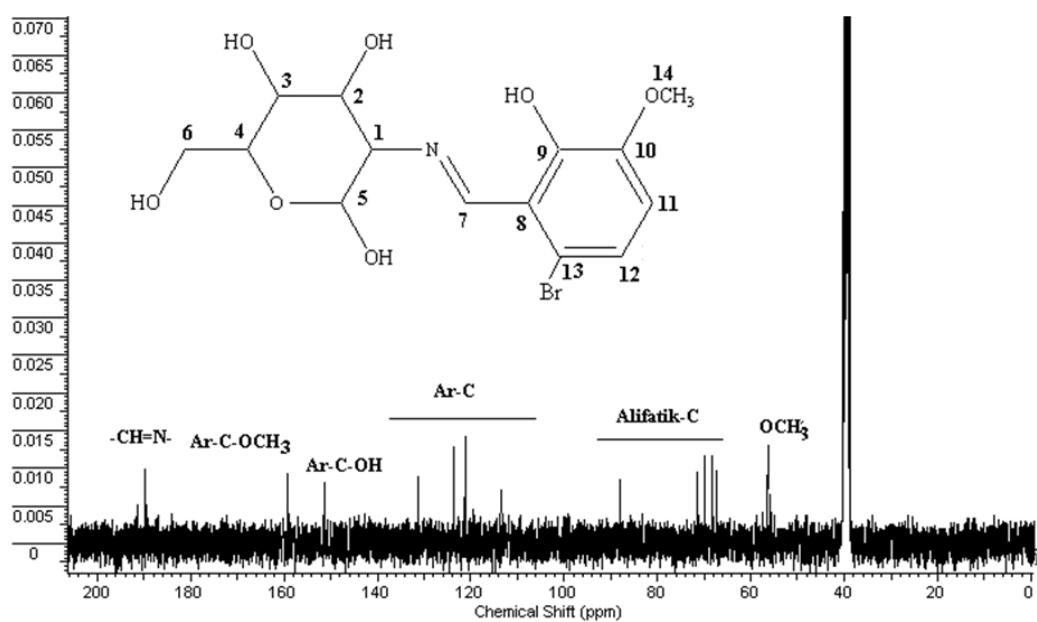


Fig. S2 ^{13}C - NMR spectrum of the compound.

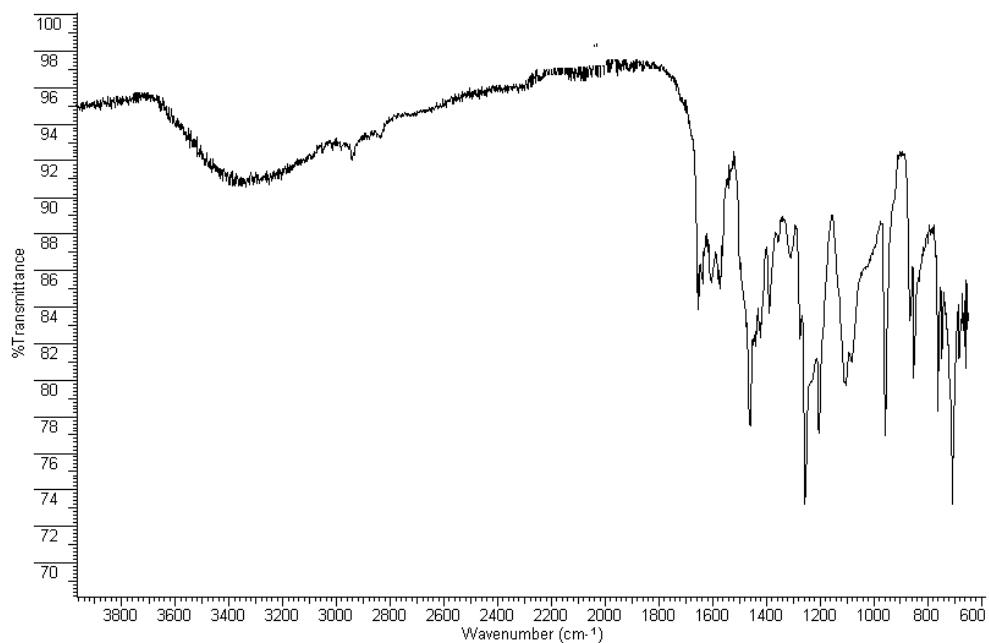


Fig. S3 FT-IR spectrum of the title compound.

Table S1 Comparison of the experimental and calculated vibrational frequencies (cm^{-1}).

Vibrational assignments (PED %)	Exp. IR	Unscaled freq.	Scaled freq.	I_{IR}
$\nu\text{O}_{30}\text{H}_{31}$ (100)	-	3850.44	3088.72	55.112
$\nu\text{O}_{30}\text{H}_{31}$ (89)	-	3848.95	3687.29	48.768
$\nu\text{O}_{30}\text{H}_{31}$ (99)	-	3800.46	3640.84	60.302
$\nu\text{O}_{30}\text{H}_{31}$ (100)	-	3784.99	3626.02	67.800
$\nu\text{C}_3\text{H}_8$ (98)	-	3223.04	3087.67	2.387
$\nu\text{O}_{11}\text{H}_{12}$ (12) + $\nu\text{C}_1\text{H}_7$ (86)	-	3192.78	3058.68	71.129
$\nu\text{O}_{11}\text{H}_{12}$ (87)	-	3189.07	3055.13	550.857
$\nu\text{C}_{14}\text{H}_{15}$ (100)	2941	3064.26	2935.56	33.668
νCH (97)	2836	3000.97	2874.93	37.934
νNC (69)	1655	1683.80	1655.18	324.103
βCCC (26)+ νCC (25)+ βHOC (18)	1605	1651.27	1623.20	24.311
νCC (42)+ βHCC (14)	1573	1602.96	1575.71	52.262
βHOC (18)+ βHCH (44)+ τHCOC (13)	-	1505.46	1479.86	31.775
νCC (13)+ βHOC (33)+ βHCH (22)	-	1499.30	1473.81	73.021
νCC (12)+ $\nu\text{O}_{11}\text{C}_5$ (19)+ βHCH (23)	-	1495.27	1469.86	169.440
βHCH (75)+ $\tau\text{ HCOC}$ (20)	1460	1493.01	1467.63	18.957
βHCC (15)+ βHCH (57)	-	1474.75	1449.68	90.017
βHOC (11)+ $\tau\text{ HCNC}$ (18)+ τHCCC (10)	1441	1460.03	1435.21	6.772
τHCNC (14)+ τHCCC (10)	1423	1455.61	1430.86	11.179
βHCO (18)+ γCCCH (16)	1391	1412.98	1388.96	13.567
$\nu\text{O}_{11}\text{C}_5$ (42)+ βCCC (11)	1259	1306.76	1284.55	140.904
ρHCH (10)+ τHCOC (56)	-	1206.98	1186.46	2.508
ρHCN (25)+ τHCOC (74)	-	1167.64	1147.79	0.749
$\nu\text{O}_{30}\text{C}_{21}$ (16)	1082	1111.18	1092.29	63.390
$\nu\text{O}_{30}\text{C}_{21}$ (33)	-	1083.06	1064.64	99.785
$\nu\text{O}_{30}\text{C}_{21}$ (40)	-	1077.14	1058.83	7.009
νNC (33)+ $\nu\text{O}_{30}\text{C}_{21}$ (10)	-	1065.85	1047.73	271.115
$\nu\text{O}_{30}\text{C}_{21}$ (18)+ $\nu\text{O}_{36}\text{C}_{24}$ (40)	-	1054.44	1036.52	89.557
$\nu\text{O}_{30}\text{C}_{21}$ (20)+ $\nu\text{O}_{36}\text{C}_{24}$ (16)	1023	1047.22	1029.42	62.326
νCC (12)+ $\nu\text{O}_{30}\text{C}_{21}$ (10)+ $\nu\text{O}_{13}\text{C}_{14}$ (26)	959	988.47	971.66	13.144
νOC (12)+ βCCC (43)	855	873.69	858.84	26.559
νCC (18)+ νOC (16)	764	783.37	770.05	29.615

Vibrational modes: ν , stretching; β , in-plane bending; δ , scissoring; γ , out of plane bending; τ , torsion; ρ , rocking. ^aPotential energy distribution (PED), less than 10% are not shown.

Table S2 The calculated quantum molecular descriptors of the compound.

Parameter	B3LYP-6-311++G(d,p)
E_{HOMO} (eV)	-6.03
E_{LUMO} (eV)	-1.94
$E_{\text{HOMO}} - E_{\text{LUMO}}$	4.09
Ionization potential [$I = -E_{\text{HOMO}}$] (eV)	6.03
Electron affinity [$A = -E_{\text{LUMO}}$] (eV)	1.94
Chemical hardness [$\eta = (I-A)/2$] (eV)	2.04
Electronegativity [$\mu = -(I+A)/2$] (eV)	3.99
Softness [$\zeta = 1/2\eta$] (eV)	0.24
Electrophilicity index [$\omega = \mu^2/2\eta$] (eV)	3.88

Table S3 The calculated results with second order perturbation theory of Fock matrix in NBO of the compound.

ED(i)(e)	Donor (i)	ED(j)(e)	Acceptor (j)	E(2) ^a kcal/mol	E(j)-E(i) ^b a.u.	F(i,j) ^c a.u.
1.75115	$\pi(\text{C1-C2})$	0.36526	$\pi^*(\text{C3-C4})$	17.36	0.29	0.066
1.75115	$\pi(\text{C1-C2})$	0.42805	$\pi^*(\text{C5-C6})$	14.53	0.29	0.061
1.96117	$\sigma(\text{C1-C6})$	0.03421	$\sigma^*(\text{C2-Br18})$	5.93	0.78	0.061
1.97706	$\sigma(\text{C1-H7})$	0.02577	$\sigma^*(\text{C2-C3})$	4.40	1.06	0.061
1.97014	$\sigma(\text{C3-C4})$	0.02577	$\sigma^*(\text{C2-C3})$	4.13	1.26	0.064
1.97014	$\sigma(\text{C3-C4})$	0.03421	$\sigma^*(\text{C2-Br18})$	4.79	0.81	0.056
1.70073	$\pi(\text{C3-C4})$	0.37007	$\pi^*(\text{C1-C2})$	18.13	0.29	0.066
1.70073	$\pi(\text{C3-C4})$	0.42805	$\pi^*(\text{C5-C6})$	17.94	0.29	0.067
1.97437	$\sigma(\text{C3-H8})$	0.02335	$\sigma^*(\text{C1-C2})$	4.23	1.13	0.062
1.62387	$\pi(\text{C5-C6})$	0.37007	$\pi^*(\text{C1-C2})$	21.98	0.28	0.070
1.62387	$\pi(\text{C5-C6})$	0.36526	$\pi^*(\text{C3-C4})$	16.94	0.28	0.062
1.62387	$\pi(\text{C5-C6})$	0.16023	$\pi^*(\text{C9-N10})$	20.05	0.27	0.069
1.97184	$\sigma(\text{C6-C9})$	0.02248	$\sigma^*(\text{N10-C20})$	4.48	1.04	0.061
1.94658	$\pi(\text{C9-N10})$	0.42805	$\pi^*(\text{C5-C6})$	6.54	0.36	0.048
1.98368	$\sigma(\text{O11-H12})$	0.03449	$\sigma^*(\text{C4-C5})$	4.97	1.25	0.071
1.97809	$\sigma(\text{C22-H29})$	0.03780	$\sigma^*(\text{C24-O36})$	4.27	0.80	0.052
1.97361	$\sigma(\text{C23-H27})$	0.02267	$\sigma^*(\text{C21-O30})$	4.51	0.80	0.054
1.97961	$\sigma(\text{C37-H38})$	0.03780	$\sigma^*(\text{C24-O36})$	5.27	0.79	0.058
1.99426	$\sigma(\text{C37-O4})$	0.00462	$\sigma^*(\text{O40-H41})$	10.74	2.91	0.158
1.86700	n(N10)	0.03920	$\sigma^*(\text{C9-H19})$	11.01	0.76	0.084
1.86700	n(N10)	0.06477	$\sigma^*(\text{O11-H12})$	23.82	0.79	0.125
1.86700	n(N10)	0.02863	$\sigma^*(\text{C20-H25})$	5.65	0.77	0.060
1.97329	n(O11)	0.03169	$\sigma^*(\text{C5-C6})$	6.69	1.11	0.077
1.80757	n(O11)	0.42805	$\pi^*(\text{C5-C6})$	37.06	0.33	0.104
1.96068	n(O13)	0.02716	$\sigma^*(\text{C3-C4})$	6.89	1.10	0.078
1.83468	n(O13)	0.36526	$\pi^*(\text{C3-C4})$	30.98	0.33	0.096
1.83468	n(O13)	0.01863	$\sigma^*(\text{C14-H15})$	5.54	0.69	0.057
1.83468	n(O13)	0.01855	$\sigma^*(\text{C14-H17})$	5.50	0.69	0.057
1.97664	n(Br18)	0.02577	$\sigma^*(\text{C2-C3})$	3.25	0.83	0.046
1.94694	n(Br18)	0.37007	$\pi^*(\text{C1-C2})$	9.29	0.31	0.052
1.94990	n(O30)	0.04153	$\sigma^*(\text{C21-C23})$	5.45	0.67	0.054
1.94990	n(O30)	0.03343	$\sigma^*(\text{C21-H26})$	7.19	0.69	0.063
1.95907	n(O32)	0.04141	$\sigma^*(\text{C23-C24})$	7.36	0.69	0.064
1.93594	n(O34)	0.03896	$\sigma^*(\text{C22-H29})$	7.27	0.70	0.064
1.93594	n(O34)	0.05460	$\sigma^*(\text{C22-O36})$	8.57	0.62	0.065
1.90014	n(O36)	0.04581	$\sigma^*(\text{C20-C22})$	4.06	0.65	0.046
1.90014	n(O36)	0.05205	$\sigma^*(\text{C22-O34})$	13.58	0.60	0.081
1.90014	n(O36)	0.04141	$\sigma^*(\text{C23-C24})$	4.09	0.67	0.047
1.90014	n(O36)	0.03564	$\sigma^*(\text{C24-C37})$	6.73	0.68	0.061
1.96307	n(O40)	0.02492	$\sigma^*(\text{C37-H38})$	5.96	0.69	0.057
1.96307	n(O40)	0.02617	$\sigma^*(\text{C37-H39})$	6.12	0.72	0.060

ED is the electron density.

^a E(2) is the energy of hyperconjugative interactions.

^b Energy difference between donor (i) and acceptor (j) NBO.

^c F(i,j) is the Fock matrix element between i and j NBO.

Non-linear optical effects

Non-linear optical materials continue to attract the attention of researchers due to their future potential applications in the optical technology and industrial applications. It is well known that the higher values of molecular polarizability and hyperpolarizability are important for more active NLO properties. To understand the NLO properties of the title compound, the linear polarizability (α) and the first hyperpolarizability (β) were calculated at the B3LYP/6-311++G(d,p) level using Gaussian 09W program package. The polarizabilities and hyperpolarizability are reported in terms of atomic units (a.u.) and the calculated values have been converted by using $1 \text{ a.u}^3 = (0.529)^3 \text{ \AA}^3$ for α and $1 \text{ a.u} = 8.641 \times 10^{-33} \text{ cm}^5/\text{esu}$ for β .³⁰ The calculated polarizability (α) and first hyperpolarizability (β) for the compound were found as 35.080 \AA^3 and $4.920 \times 10^{-30} \text{ cm}^5/\text{esu}$, respectively (Table S4). Urea is one of the essential molecules used for determination of the NLO properties of molecular systems. Therefore, it is usually used as reference molecule in NLO studies. The calculated values of urea at the B3LYP/6-311++G(d,p) level were found as 4.90 \AA^3 for α and $0.781 \times 10^{-30} \text{ cm}^5/\text{esu}$ for β . The polarizability and first hyperpolarizability for title molecule is approximately 8.5 and 8-fold larger than those of urea, which implies that the compound is a good candidate material for NLO applications.

Table S4 The electric dipole moment, polarizability and first order hyperpolarizability values of the compound.

Parameters	Value (a.u.)	Parameters	Value (a.u.)
α_{xx}	326.9116171	β_{xxx}	38.7367578
α_{xy}	-0.9075499	β_{xxy}	-143.5708925
α_{xz}	0.5716699	β_{xxy}	-65.0675367
α_{yy}	226.3044961	β_{yyy}	-377.2992885
α_{yz}	-0.0598852	β_{xxz}	-36.6627469
α_{zz}	157.8739912	β_{yyz}	-15.3591404
α_{total}	711.0901044	β_{xzz}	110.2477135
$\Delta\alpha$	237.0300348	β_{yzz}	-32.0047425
μ_x	-0.4707	β_{zzz}	-55.3257251
μ_y	1.8490	β_0	569.417458
μ_z	0.0637		
μ_{total}	1.9090		

Table S5 MIC ($\mu\text{g/mL}$) of the compound.

Microorganisms	Compound	Gentamicin	Ampicillin	Fluconasol
<i>S. aureus ATCC 25923</i>	128	1.25	0.50	-
<i>E. faecalis ATCC 29212</i>	32	20	8	-
<i>B. subtilis ATCC 6633</i>	64	-	2	-
<i>E. coli ATCC 25922</i>	256	2.50	4	-
<i>E. coli NRRL B-3704</i>	128	0.67	4	-
<i>P. aeruginosa ATCC 254992</i>	-	-	0.5	-
<i>P. vulgaris ATCC 13315</i>	128	2.50	8	-
<i>C. albicans ATCC 60193</i>	2	-	-	64
<i>C. tropicalis ATCC 13803</i>	64	-	-	128